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#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(\text{C-C}) = 0.003 \text{ Å}$  R factor = 0.040 wR factor = 0.114Data-to-parameter ratio = 14.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 3-[(2,4-Dinitrophenyl)hydrazonomethyl]phenol dimethylformamide solvate

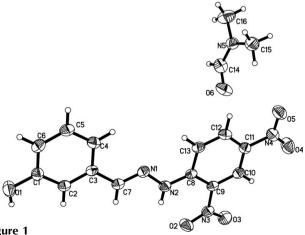
In the title compound,  $C_{13}H_{10}N_4O_5\cdot C_3H_7NO$ , the Schiff base is approximately planar. Molecules are connected via weak intermolecular  $O-H\cdots O$ ,  $N-H\cdots O$  and  $C-H\cdots O$  hydrogen bonds.

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### Comment

Metal complexes based on Schiff bases have attracted much attention because they can be utilized as model compounds of active centres in various proteins and enzymes (Kahwa *et al.*, 1986; Santos *et al.*, 2001). As part of an investigation of the coordination properties of Schiff bases functioning as ligands (Yu *et al.*, 2005; Deng *et al.*, 2005; Jing, Fan *et al.*, 2005*a,b*; Jing, Wang *et al.*, 2005; Jing *et al.*, 2006), we report here the synthesis and crystal structure of the title compound, (I).

The asymmetric unit of (I) comprises one molecule each of 3-[(2,4-dinitrophenyl)hydrazonomethyl]phenol and dimethyl-



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The structure of (I), showing 30% probability displacement ellipsoids.

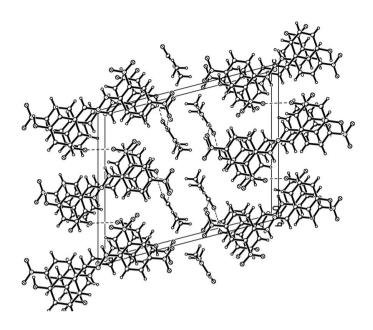


Figure 2 The packing of (I), viewed down the b axis, showing intermolecular hydrogen bonds (dashed lines).

formamide (Fig. 1). The Schiff base is approximately planar; the central chromophore (C8–C13/N1–N4) and the 3-hydroxybenzaldehyde group (C1–C7/O1) are each planar, with r.m.s. deviations of 0.020 and 0.008 Å, respectively. The dihedral angle between these planes is  $2.44 (3)^{\circ}$ .

An intramolecular N $-H\cdots O$  hydrogen bond stabilizes the molecular structure, while intermolecular  $O-H\cdots O$ , N $-H\cdots O$  and  $C-H\cdots O$  hydrogen bonds stabilize the crystal structure (Table 2 and Fig. 2). Inversion-related Schiff base molecules are linked  $via\ O-H\cdots O$  and  $C-H\cdots O$  hydrogen bonds involving the dimethylformamide molecules. A short  $O2\cdots O2(-x,y,\frac{1}{2}-z)$  contact of 2.750 (2) Å is also observed in the crystal structure.

## **Experimental**

An anhydrous ethanol solution (50 ml) of 3-hydroxybenzaldehyde (1.22 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 2,4-dinitrophenylhydrazine (2.03 g, 10 mmol), and the mixture was stirred at 330 K for 6 h under  $N_2$ , whereupon a red precipitate appeared. The product was isolated, recrystallized from  $N_2$ ,  $N_3$ -dimethylformamide and then dried in vacuo to give pure compound (I) in 89% yield. Red single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an  $N_3$ -dimethylformamide solution.

## Crystal data

 $C_{13}H_{10}N_4O_5 \cdot C_3H_7NO$  $D_x = 1.402 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation  $M_r = 375.35$ Monoclinic, P2/c Cell parameters from 2708 a = 17.701 (4) Å reflections b = 7.0661 (18) Å $\theta = 2.8 – 25.7^{\circ}$  $\mu = 0.11 \text{ mm}^{-1}$ c = 14.695 (4) Å  $\beta = 104.607 (4)^{\circ}$ T = 294 (2) K $V = 1778.6 (8) \text{ Å}^3$ Block, red Z = 4 $0.40 \times 0.32 \times 0.20 \text{ mm}$ 

### Data collection

Bruker SMART CCD area-detector diffractometer 2221 reflections with  $I > 2\sigma(I)$   $\varphi$  and  $\omega$  scans  $R_{\rm int} = 0.030$  Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  $m_{\rm max} = 26.3^{\circ}$   $m_{\rm min} = 0.943, T_{\rm max} = 0.978$   $m_{\rm max} = 0.978$   $m_$ 

#### Refinement

Refinement on  $F^2$  $w = 1/[\sigma^2(F_0^2) + (0.0478P)^2]$  $R[F^2 > 2\sigma(F^2)] = 0.040$ + 0.3148P]  $wR(F^2) = 0.114$ where  $P = (F_0^2 + 2F_c^2)/3$ S = 1.00 $(\Delta/\sigma)_{\rm max} = 0.001$ 3634 reflections  $\Delta \rho_{\text{max}} = 0.16 \text{ e Å}$  $\Delta \rho_{\min} = -0.14 \text{ e Å}^{-3}$ 255 parameters H atoms treated by a mixture of Extinction correction: SHELXL97 independent and constrained (Sheldrick, 1997) refinement Extinction coefficient: 0.0309 (18)

**Table 1** Selected geometric parameters (Å, °).

N1-C7	1.271 (2)	C3-C4	1.394 (2)
N1-N2	1.3730 (19)	C3-C7	1.457 (2)
N2-C8	1.345 (2)	C8-C13	1.413 (2)
N3-O2	1.2276 (18)	C8-C9	1.419 (2)
N3-C9	1.445 (2)		, ,
C7-N1-N2	116.42 (15)	N1-C7-C3	121.61 (16)
C8-N2-N1	119.53 (15)	N2-C8-C13	120.27 (15)
O2-N3-C9	119.34 (14)	N2-C8-C9	123.72 (15)
C4-C3-C7	122.23 (16)	C8-C9-N3	121.77 (15)

**Table 2** Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdot\cdot\cdot A$
$O1-H1\cdots O6^{i}$ $N2-H2A\cdots O2^{ii}$ $N2-H2A\cdots O2$ $C14-H14\cdots O5^{iii}$	0.92 (3)	1.73 (3)	2.634 (2)	168 (2)
	0.855 (18)	2.586 (18)	3.385 (2)	156.1 (15)
	0.855 (18)	2.009 (18)	2.624 (2)	128.1 (15)
	0.93	2.39	3.271 (3)	158

Symmetry codes: (i) -x, -y + 1, -z + 1; (ii) -x, y,  $-z + \frac{1}{2}$ ; (iii) x, y + 1, z.

H atoms attached to N and O atoms were located in a difference Fourier map and refined freely. H atoms bonded to C atoms were included in calculated positions [C—H = 0.93 (aromatic) or 0.96 Å (methyl)] and refined using a riding-model approximation, with  $U_{\rm iso}({\rm H})=1.2 U_{\rm eq}({\rm C})$  or  $1.5 U_{\rm eq}({\rm methyl})$  C).

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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